

# Chain Shuffling: A Synthetic Pathway to Multiblock Copolymers

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Abhishek Banerjee



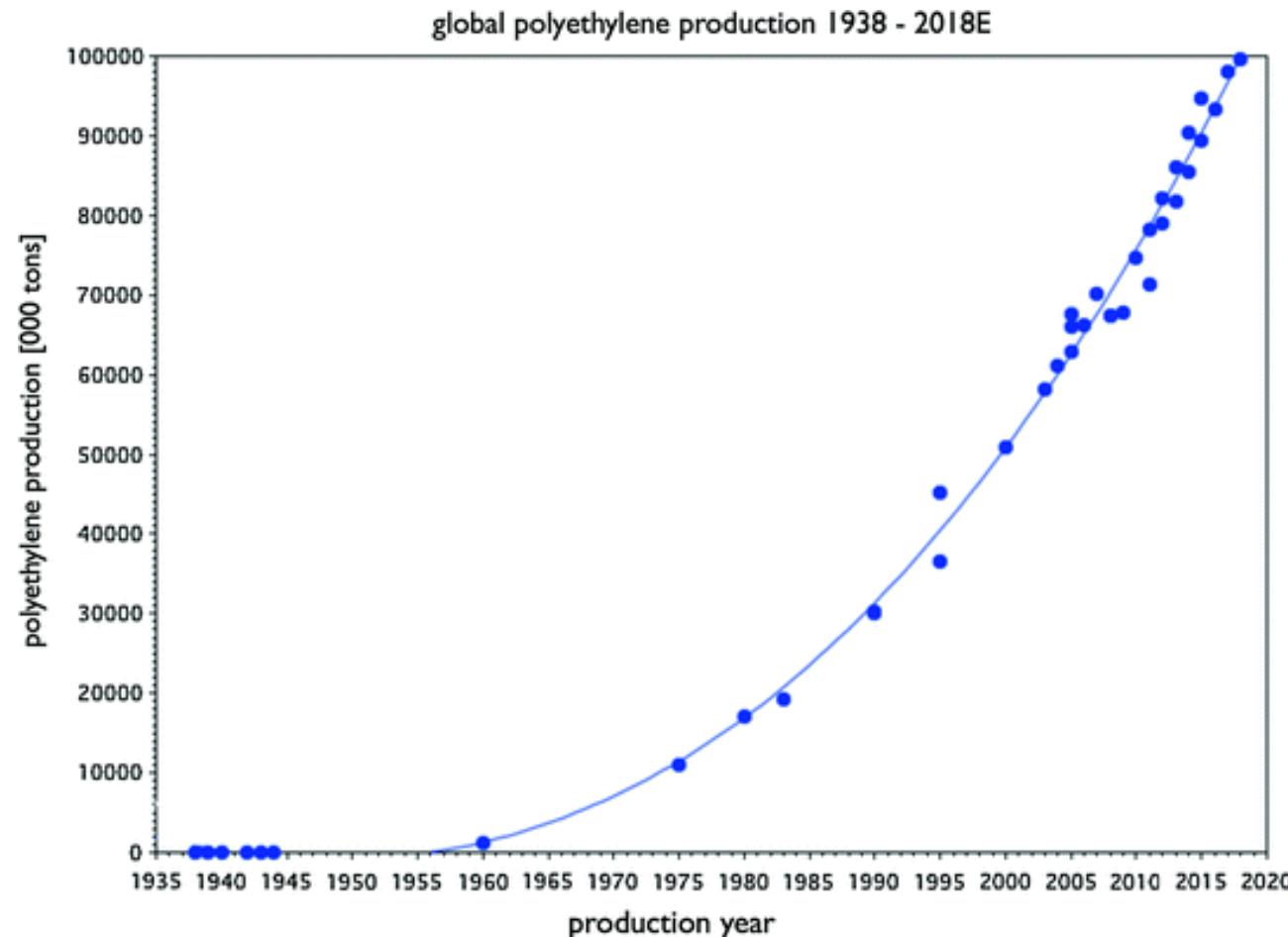
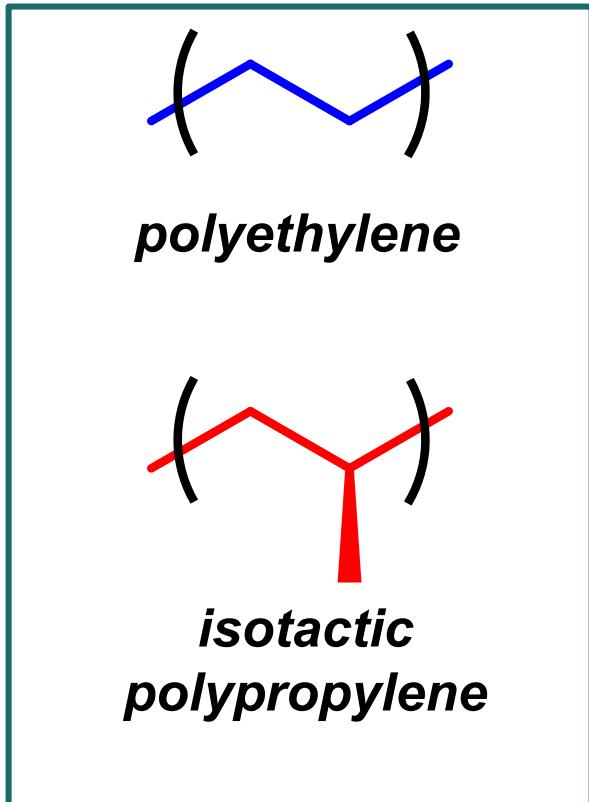
[www.eaganlab.com](http://www.eaganlab.com)

Tuesday, April 18, 2023

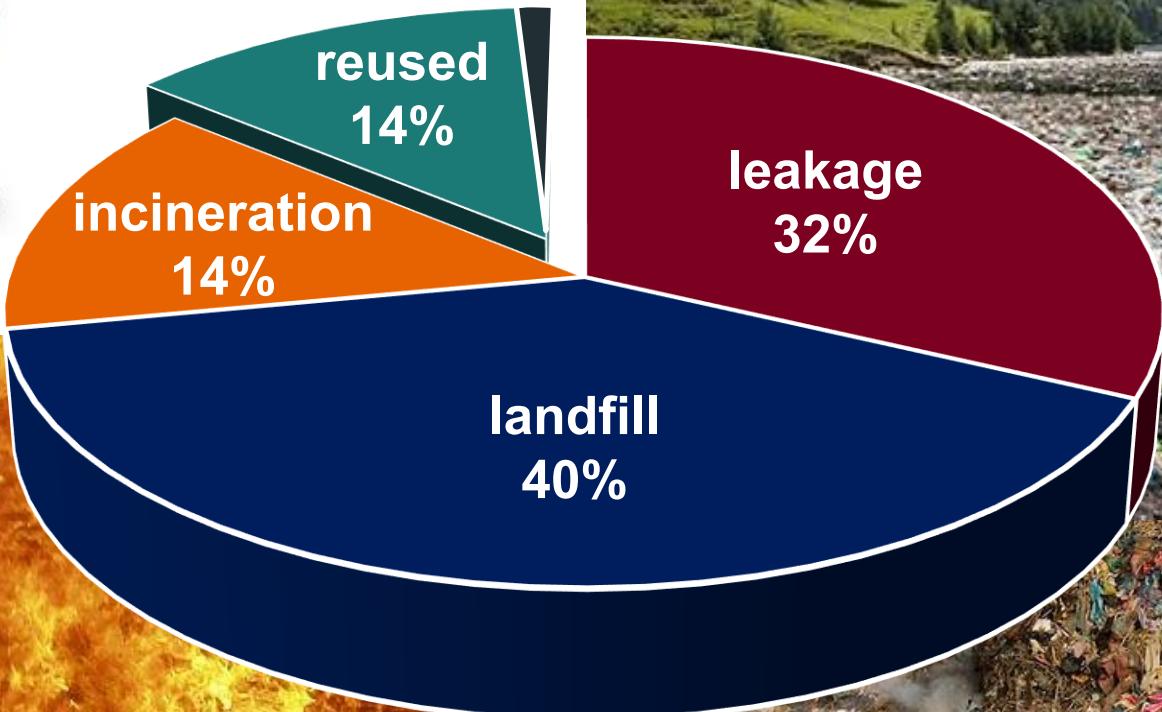


# *Abundance of PE/iPP*

**top two polymers**



# Recycling

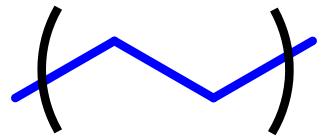


Ellen Macarthur Foundation, World Economic Forum, McKinsey & Co.  
*The New Plastics Economy* (2016).

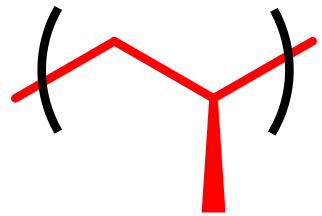
# *The Challenge*

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*top two polymers*



*polyethylene*



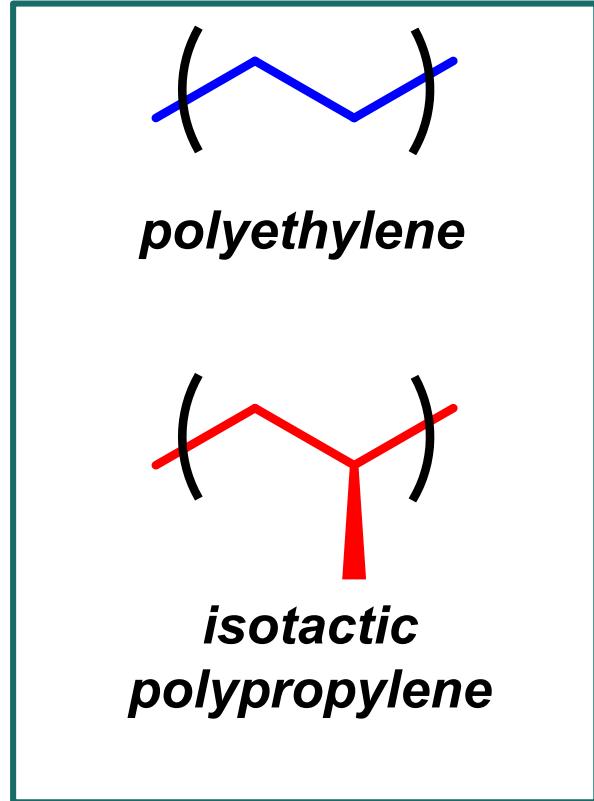
*isotactic  
polypropylene*

*over 120 million tons  
produced annually*

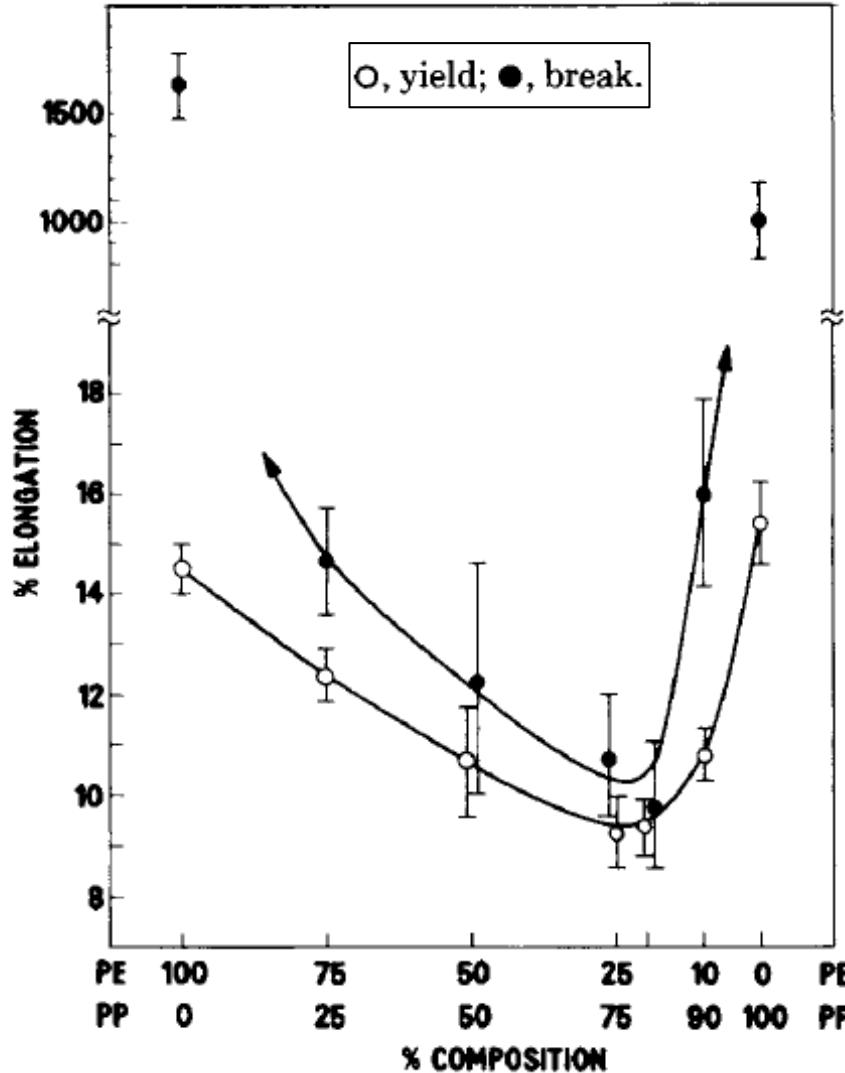


# The Challenge

*top two polymers*



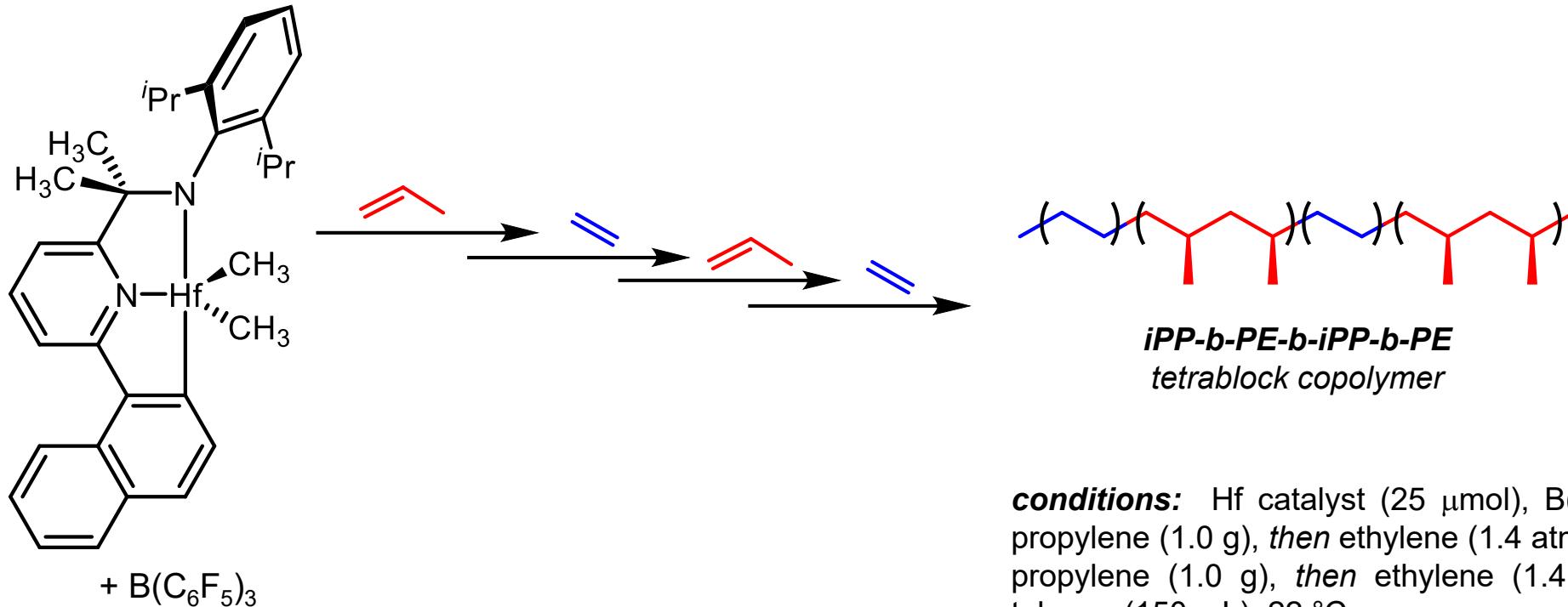
*over 120 million tons  
produced annually*



- Virgin PE and virgin iPP have high % elongations
- The addition of even a slight minor component drastically reduces % elongation

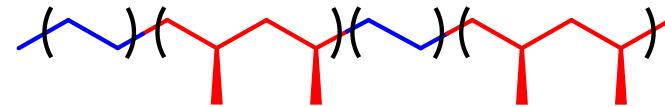
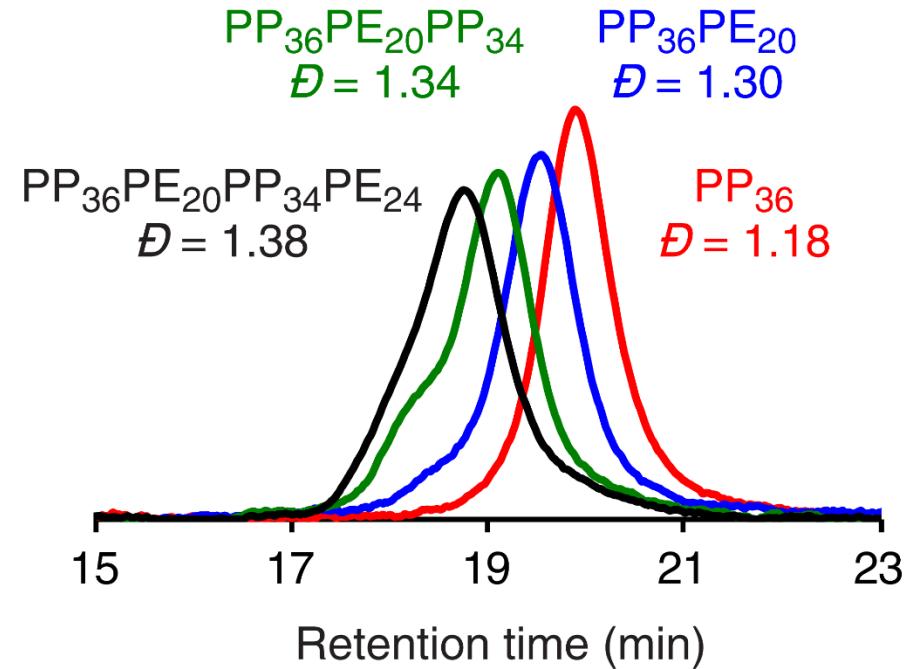
# Tetrablock Copolymers

Pyridyl amidohafnium catalysts enable  
multiblock architectures



**conditions:** Hf catalyst (25  $\mu\text{mol}$ ),  $B(C_6F_5)_3$  (25  $\mu\text{mol}$ ), propylene (1.0 g), *then* ethylene (1.4 atm), 4 minutes, *then* propylene (1.0 g), *then* ethylene (1.4 atm), 4 minutes, toluene (150 mL), 22  $^{\circ}\text{C}$ .

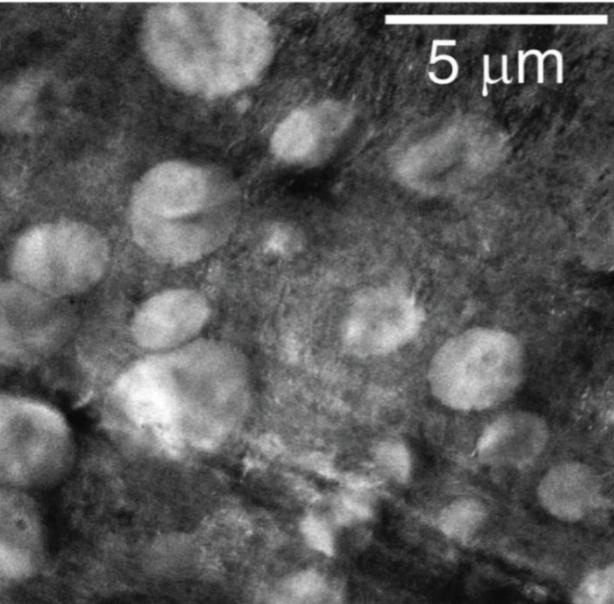
# Tetrablock Copolymers



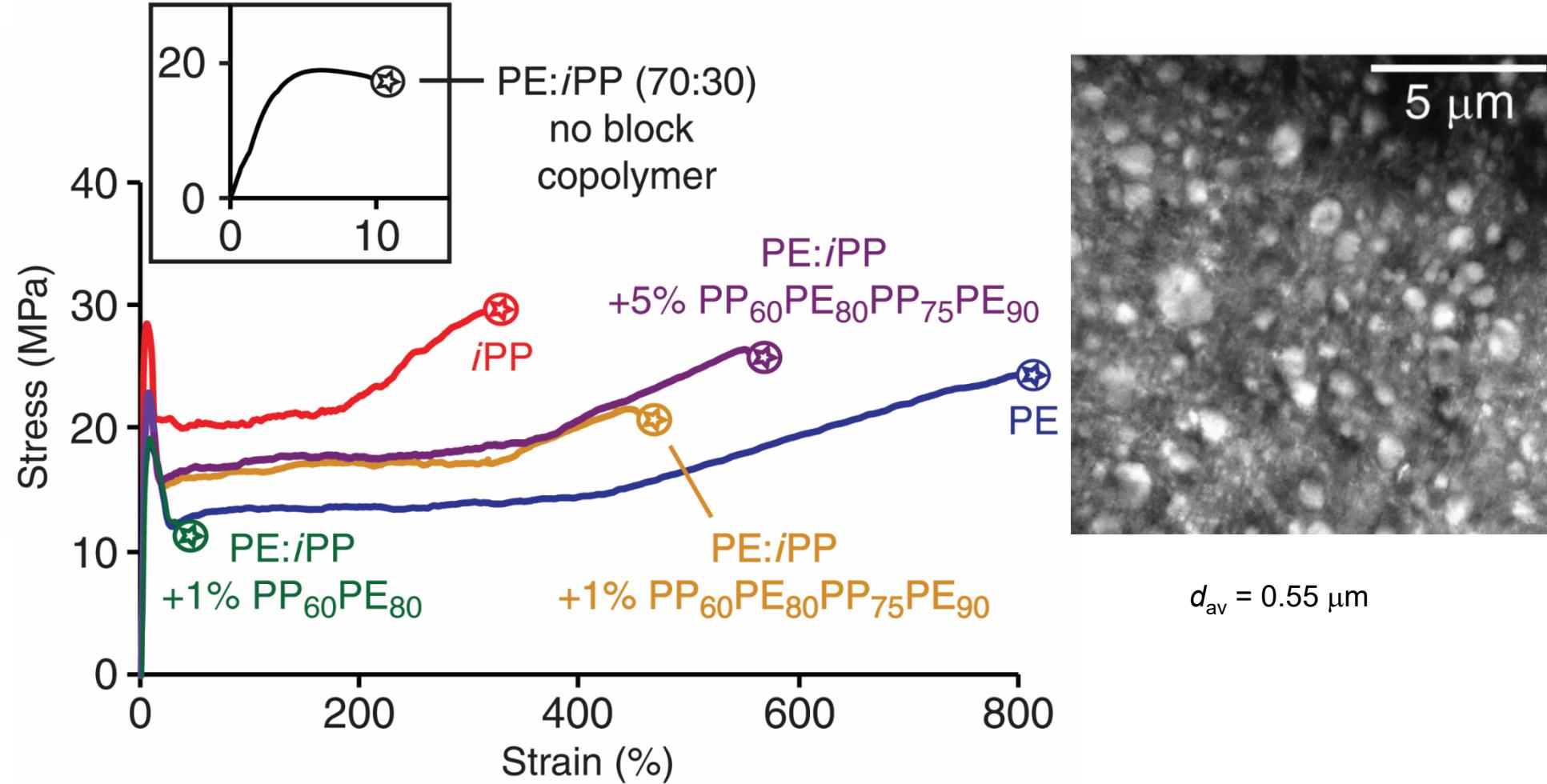
*iPP-b-PE-b-iPP-b-PE*  
tetrablock copolymer

**conditions:** Hf catalyst (25  $\mu\text{mol}$ ),  $\text{B}(\text{C}_6\text{F}_5)_3$  (25  $\mu\text{mol}$ ), propylene (1.0 g), *then* ethylene (1.4 atm), 4 minutes, *then* propylene (1.0 g), *then* ethylene (1.4 atm), 4 minutes, toluene (150 mL), 22  $^\circ\text{C}$ .

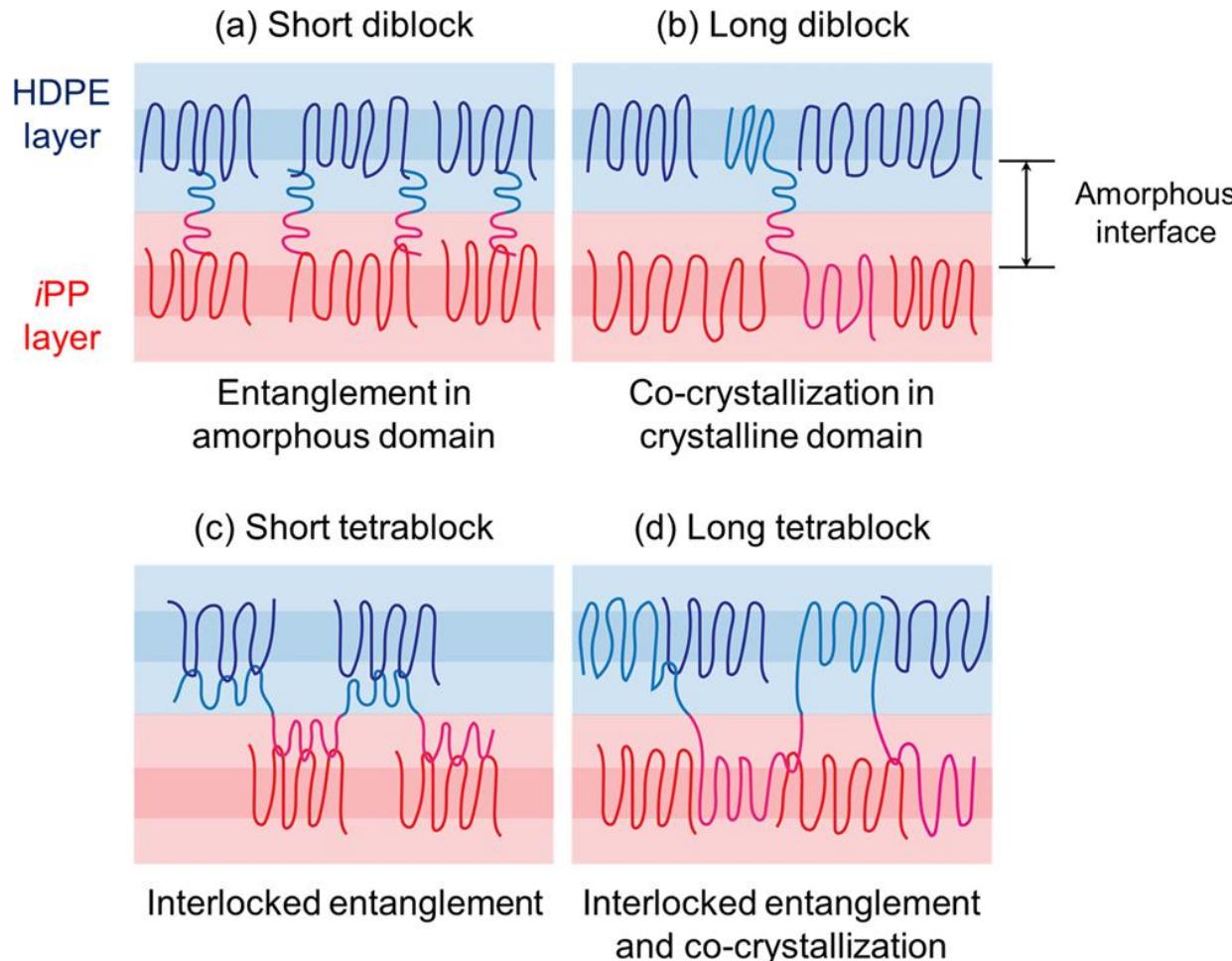
# Tensile Properties



$d_{av} = 2.16 \mu\text{m}$



# Advantages of Multi-blocks



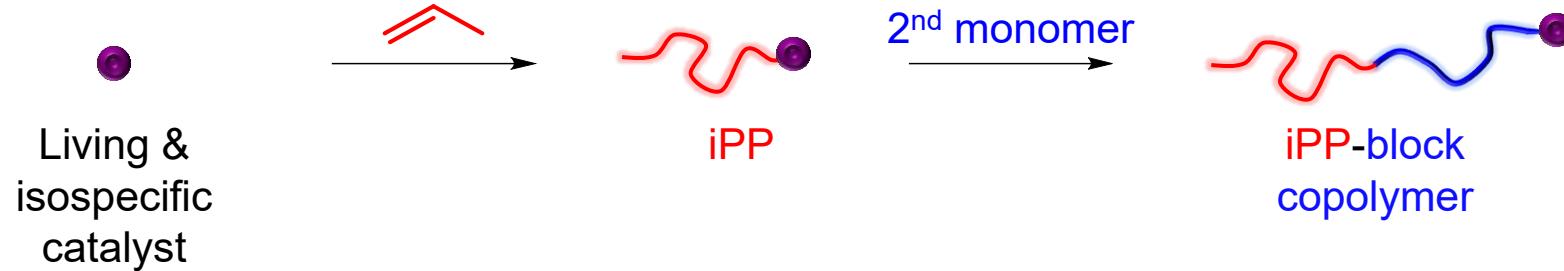
## Diblocks:

- Long diblocks co-crystallize in the crystalline domain
- Entanglements are relatively easy to dissociate

## Tetrablocks:

- Tetrablocks exhibit trapped entanglements
- Dissociating entanglements requires greater force than bulk failure of the PE matrix

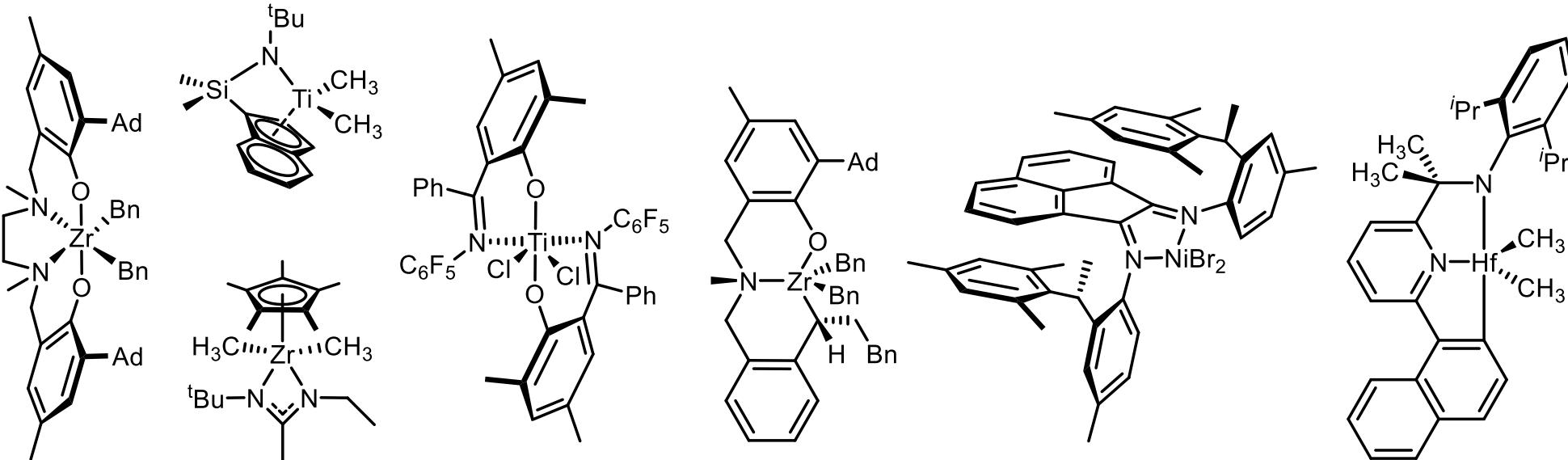
# Living Catalysts for iPP-BCP



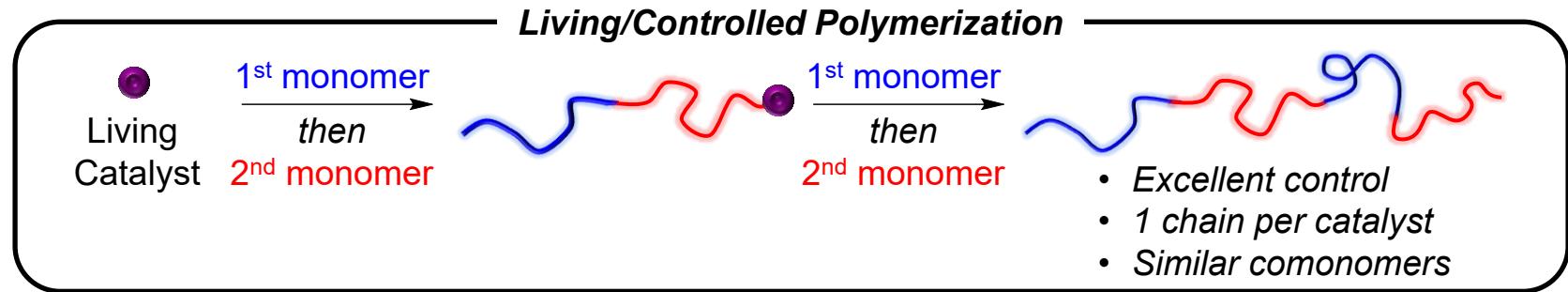
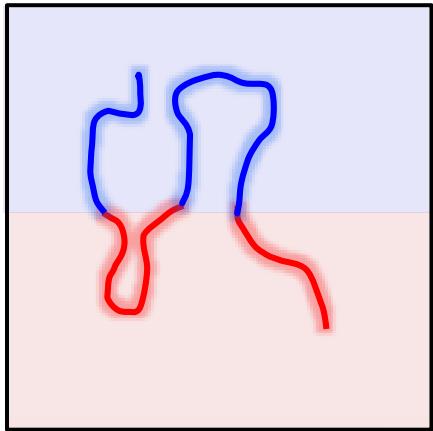
## Living Polymerization

- Excellent MW control
- Narrow dispersities
- Tacticity is challenging
- Comonomers are limited
- Not scalable

## *Examples of Living & Isospecific Catalysts*



# Limits of the System



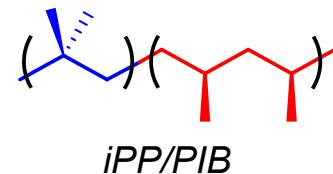
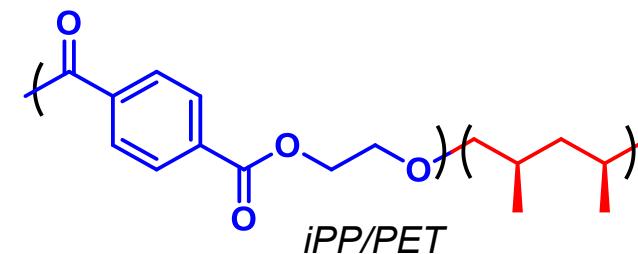
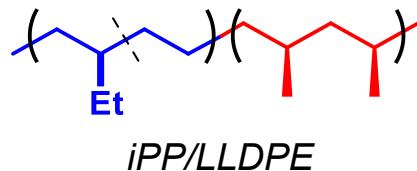
## 1) Too much catalyst

$100m \text{ tons PE}$	$2,000 \text{ tons MBCP}$	$2 \text{ mol Hf}$
$1 \text{ year}$	$1 \text{ ton PE}$	$1,000 \text{ kg MBCP}$
(annual PE)	(0.2 wt%)	(500 kDa)

$$= 400 \text{ mol Hf/year} = 71 \text{ tons Hf/year}$$

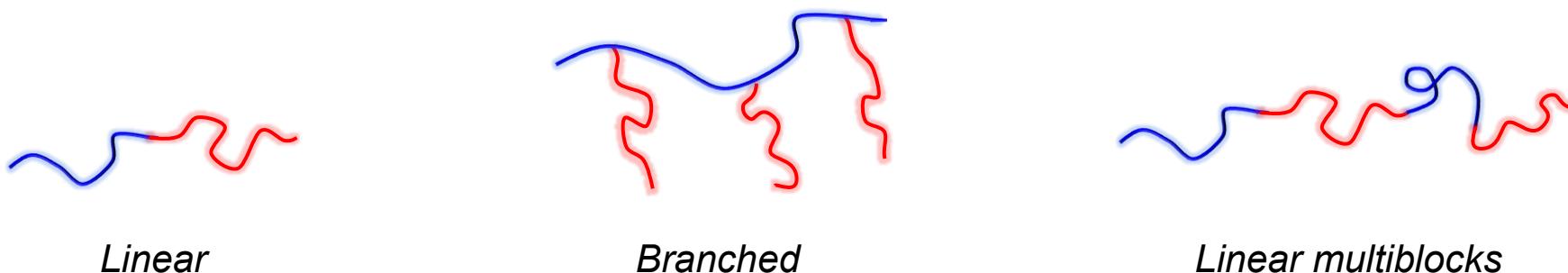
**Global production = 70 tons Hf/year**

## 2) Compositionally limited



# Polypropylene Block Copolymers

## Architecture

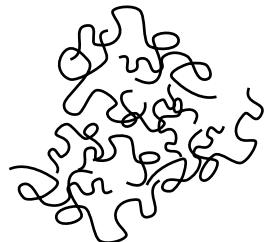
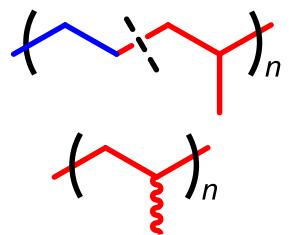


Linear

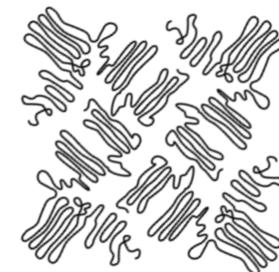
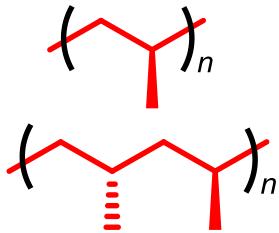
Branched

*Linear multiblocks*

## Crystallinity/Modulus

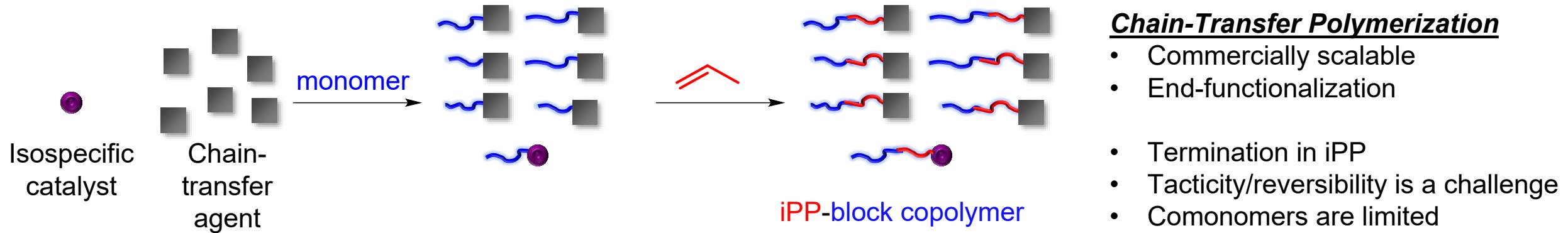


*elastomeric polypropylenes*



*crystalline polypropylenes*

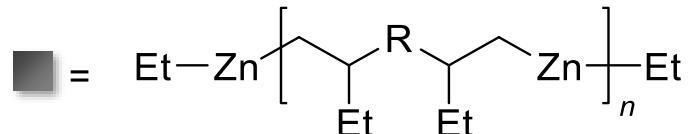
# Non-living CCTP



## Selected Examples of CCTP (iPP)



M = Zn-iPP  
B-(iPP)<sub>2</sub>  
SiH<sub>2</sub>Ph



Multifunctional CTAs enable telechelic *polyethylenes*

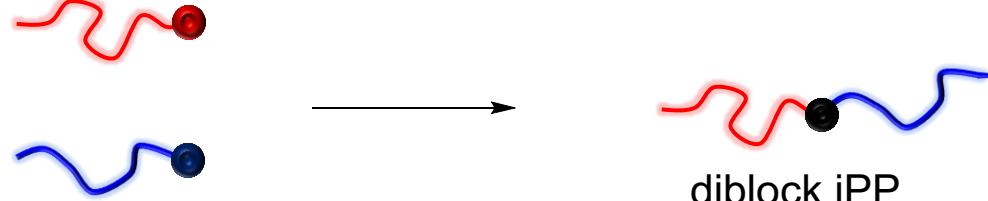


### Additional PP-CCTP references:

- Science, 2006, 312, 714.  
J. Am. Chem. Soc. 2008, 130, 442.  
Macromolecules 2000, 33, 9192.  
Macromolecules 2007, 40, 7736.  
J. Am. Chem. Soc. 2004, 126, 10701.  
Macromol. Chem. Phys. 2006, 207, 295.  
ACS Catal. 2021, 11, 10170.

# Macromolecular Engineering with iPP

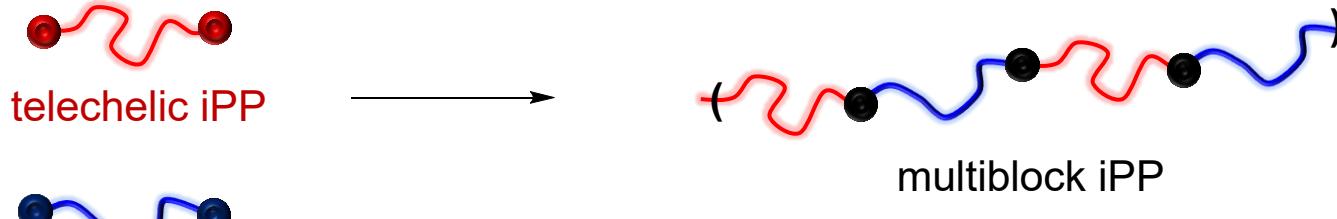
## Diblock coupling



## End-Functionalized iPP

- Non-living method
- Versatile comonomers
- Independent tuning of crystallinity

## Multiblock Step-Growth

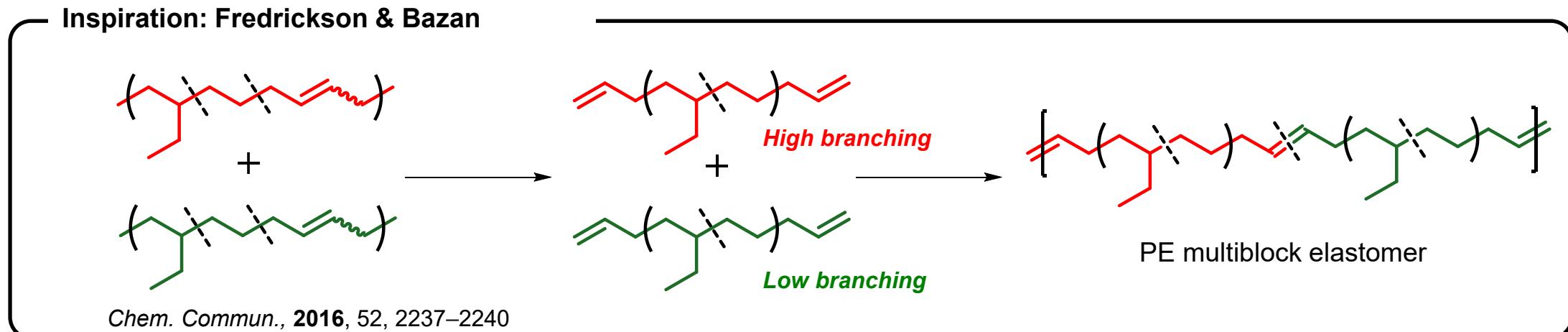
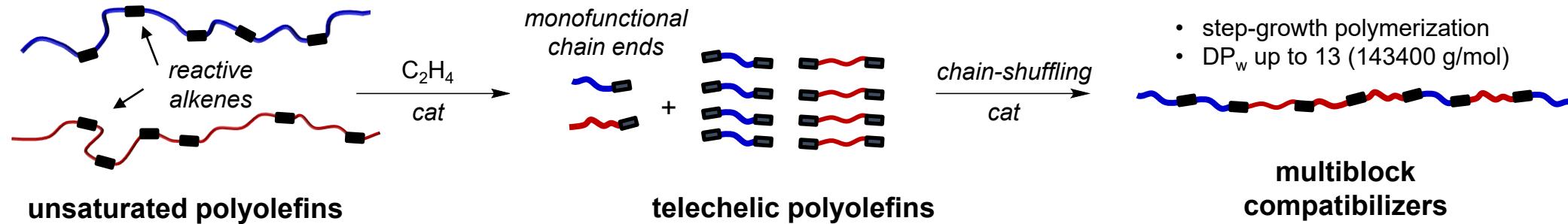


## *This work*

### Telechelic iPP

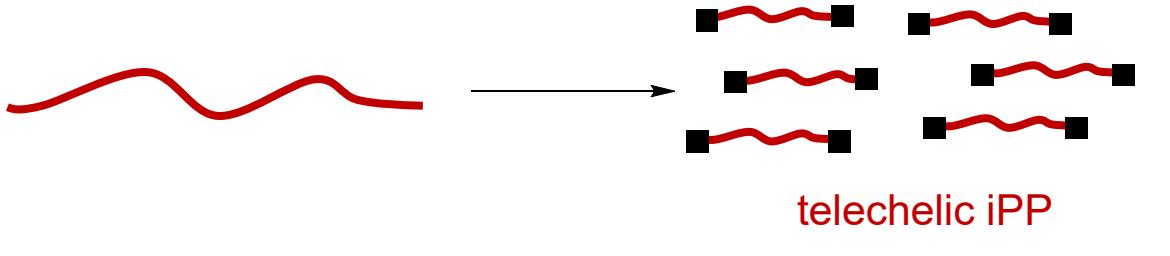
- Architecture control
- Difficult to maintain  $T_m$ ,  $M_w$ , and scalability

# Reactive Hydrocarbon Approach

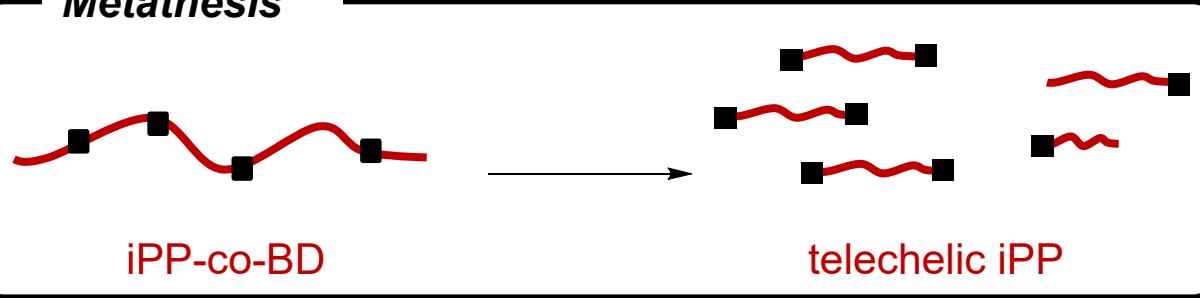


# Methods for Telechelic iPP

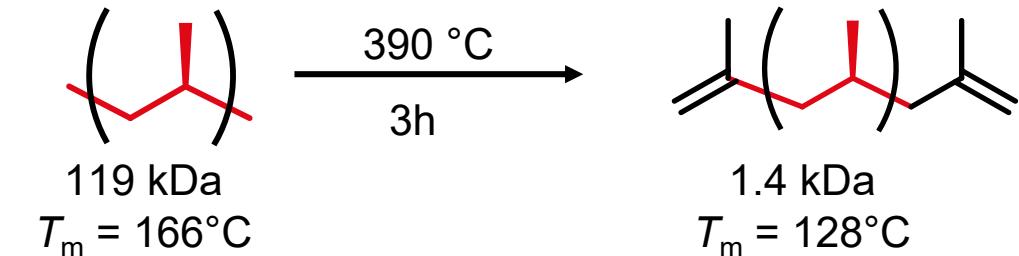
## Degradation



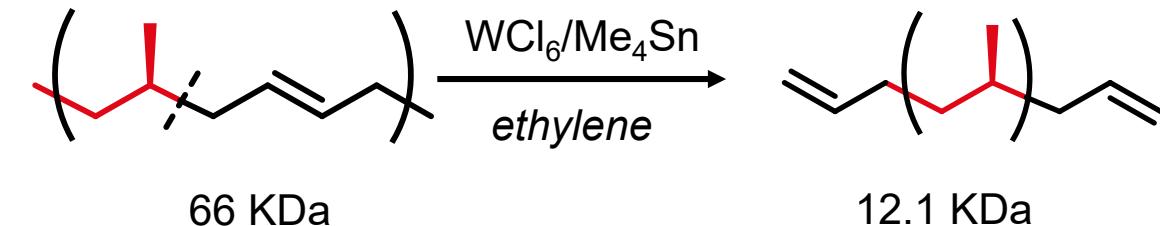
## Metathesis



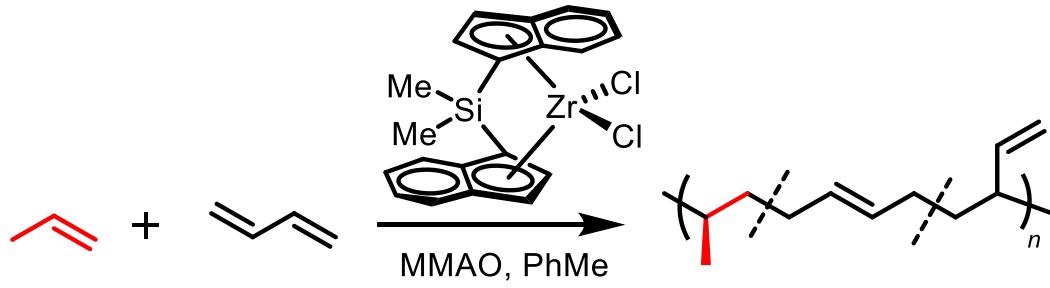
Sawaguchi and coworkers  
*Macromolecules* **1995**, *28*, 7973–7978.



Shiono and coworkers  
*Macromolecules* **2003**, *36*, 9675–9677.

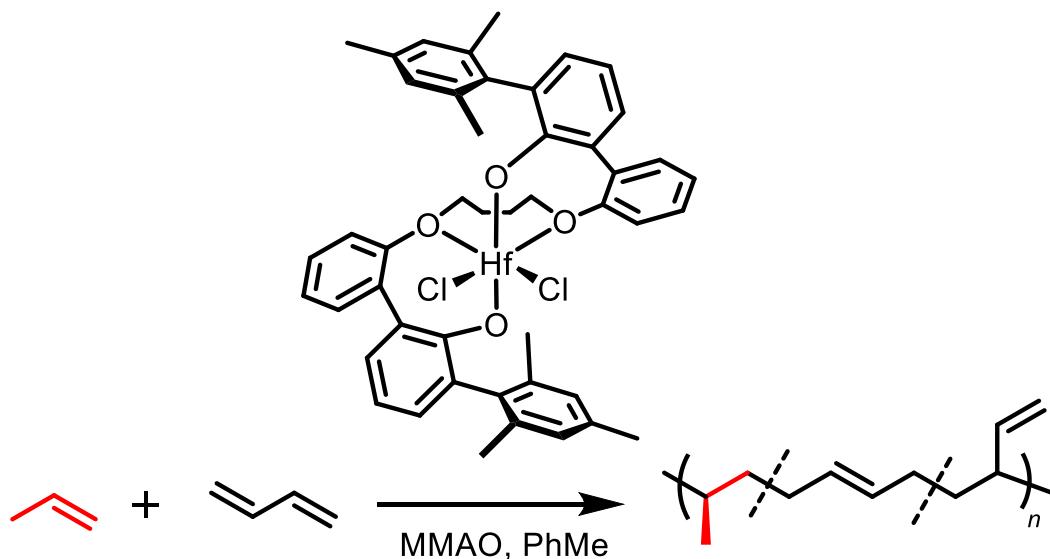


# Synthesis of Unsaturated iPP



Shiono and coworkers  
*Macromolecules* **2003**, *36*, 9675–9677.

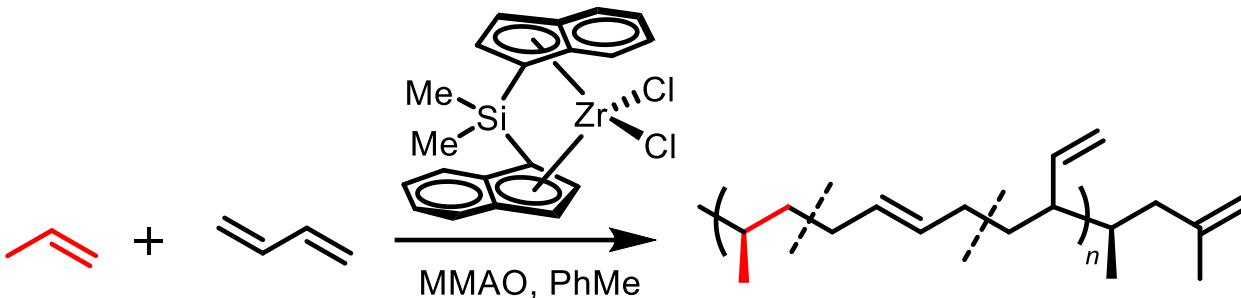
- Butadiene slows polymerization
- Yields < 2.5g
- $T_m = 97 \text{ }^\circ\text{C} — 150 \text{ }^\circ\text{C}$
- %1,4 = 0.23% — 5.38%



Coates and coworkers  
*J. Am. Chem. Soc.* **2022**, *144*, 12613–12618.

- Cyclization of dienes
- $T_m < 100 \text{ }^\circ\text{C}$
- %1,4 = 0.26% — 1.18%

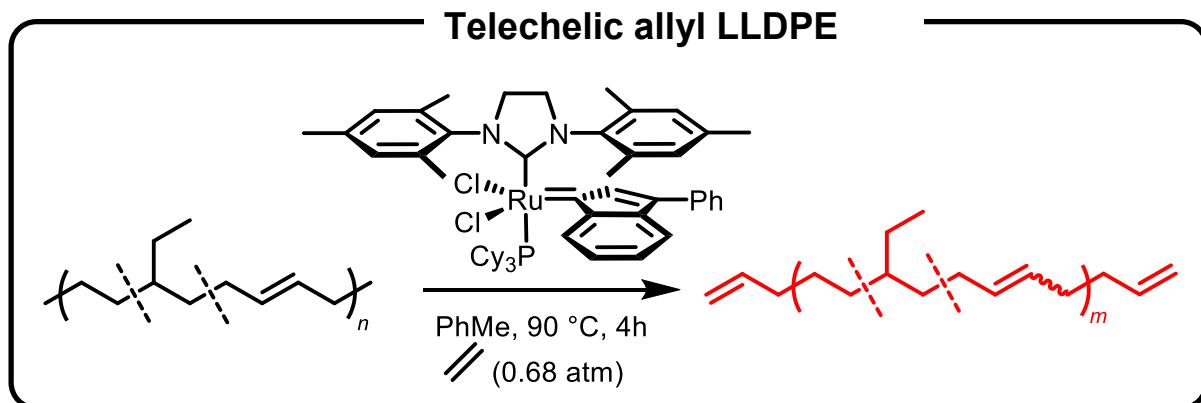
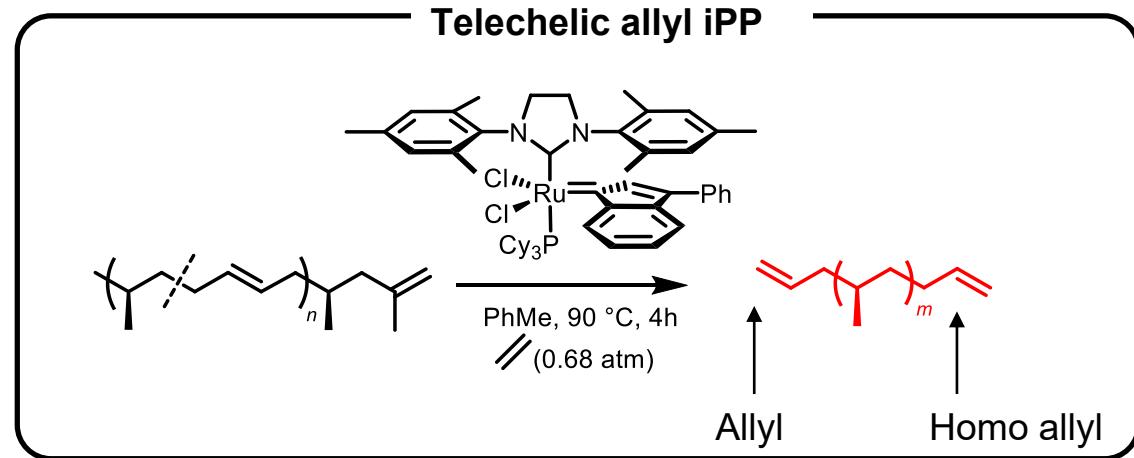
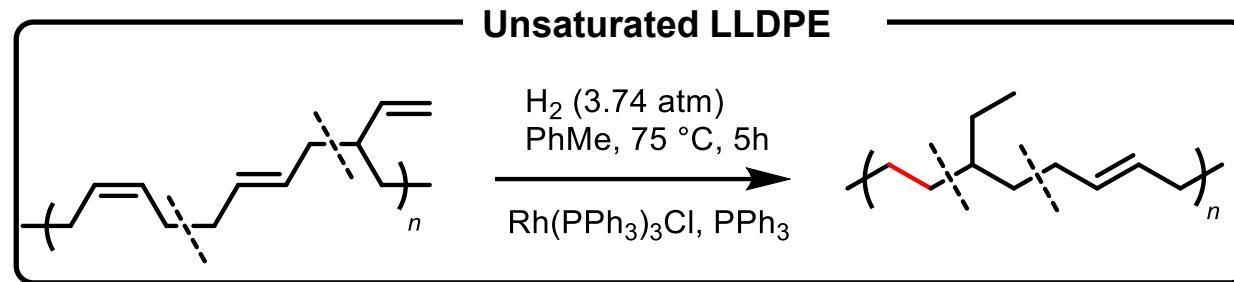
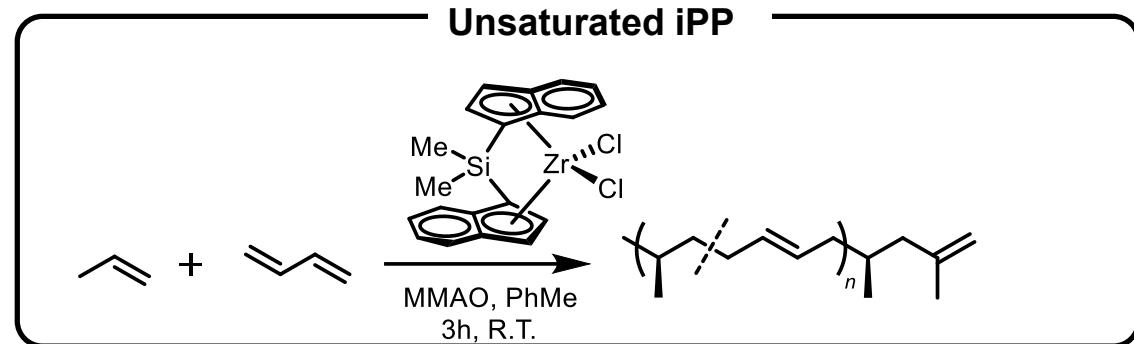
# Scalable Unsaturated iPP Synthesis



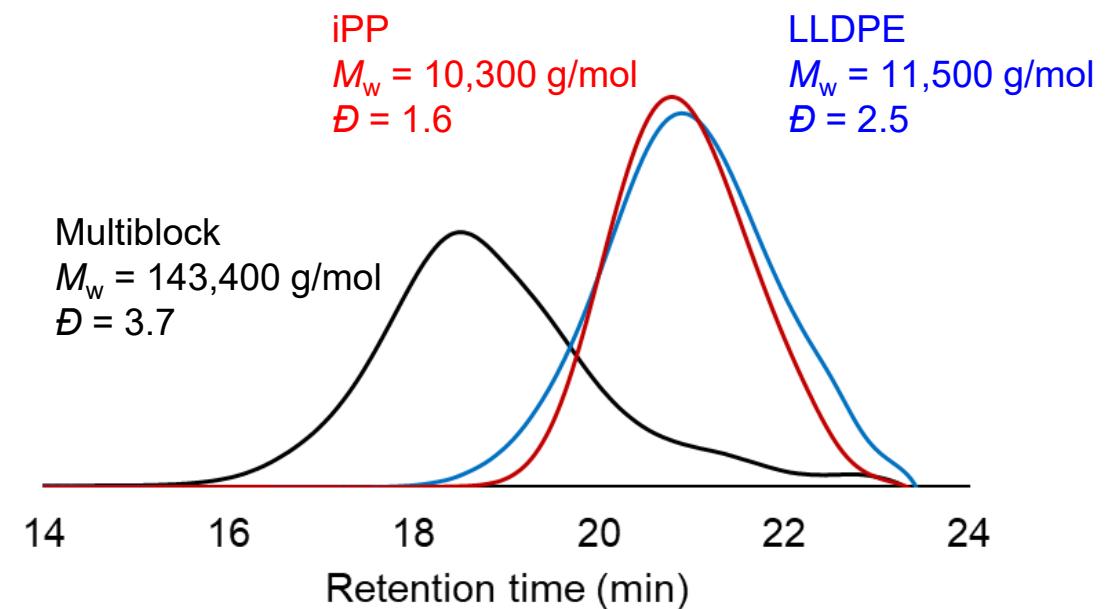
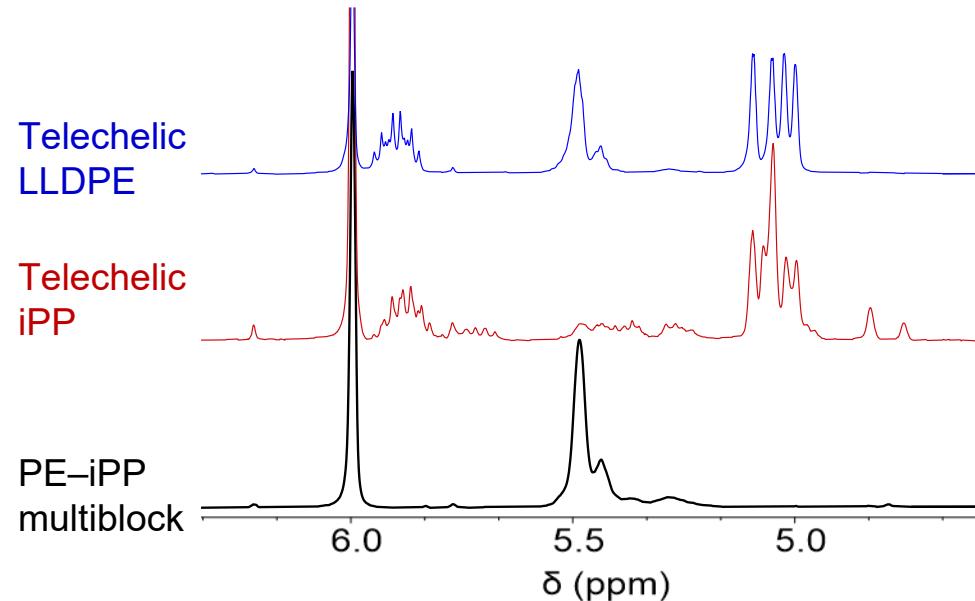
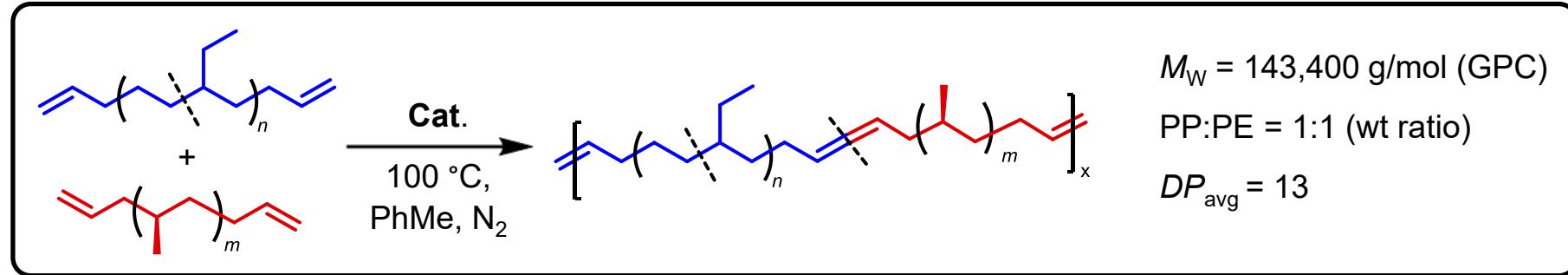
Entry (#)	Cat. conc. ( $\mu\text{mol}$ )	Initial $\text{C}_3\text{H}_6$ (mmol)	Final $\text{C}_3\text{H}_6$ (mmol)	Initial BD (mmol)	Aluminoxane (mmol)	$T_{\text{rxn}}$ ( $^{\circ}\text{C}$ )	Time (h)	Yield (g)	$M_n$ (kg/mol)	#1,2 BD (per chain)	#1,4 BD (per chain)	$T_m$ ( $^{\circ}\text{C}$ )
1	10	118.8	286.6	38.3	MMAO(1000)	25	6	2.28	26.4	1.63	6.07	139.3
2	20	118.8	413.0	74.0	MMAO(1000)	25	3	1.95	35.2	4.99	13.15	128.0
3	20	393.8	539.5	79.3	PMAO(4000)	25	3	6.13	31.3	3.10	10.83	130.9
4	20	558.4	665.3	74.5	PMAO(4000)	0	3	4.50	49.7	2.06	4.68	148.2

Nouryon's PMAO-IP improves the productivity of propene/1,3-butadiene polymerization using conventional isospecific metallocenes

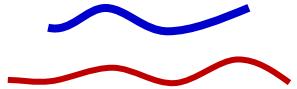
# Scalable Telechelic Polyolefin Synthesis



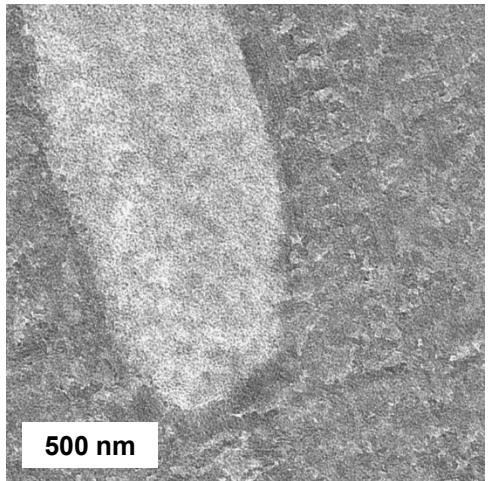
# LLDPE-ran-iPP Multiblock Copolymer



# Morphology and Crystallization



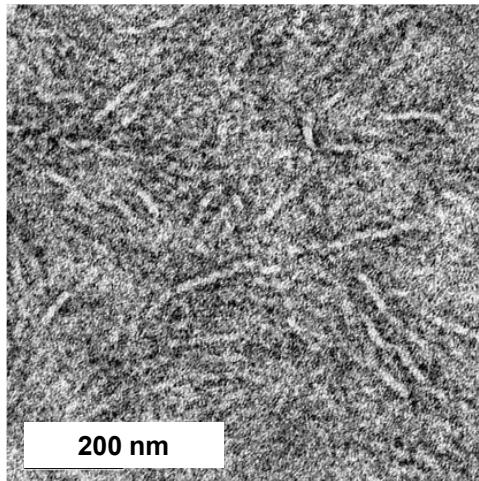
Commercial LLDPE:iPP blend



- Macro phase separation
- High crystallinity / modulus
- Poor mechanical properties



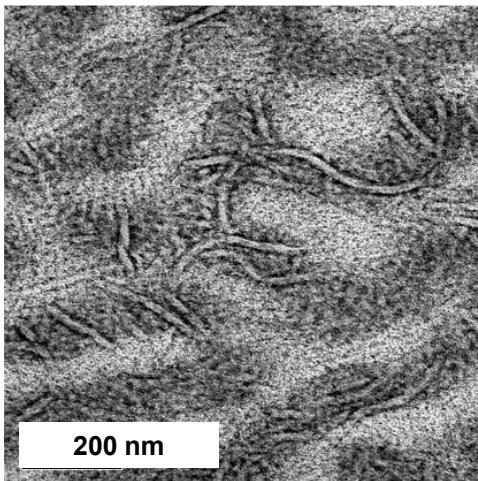
Dow Infuse OBC (multiblocks)



- Homogeneous phase
- Low crystallinity / modulus
- Elastomeric properties



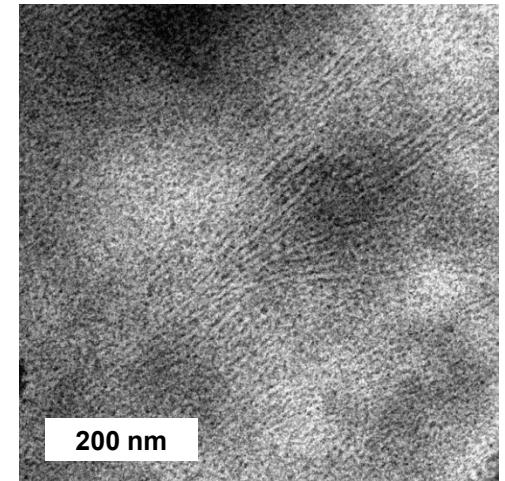
Dow INTUNE (PE-PP diblocks)



- Micro phase separation
- High crystallinity / modulus
- Compatibilizer



Multiblock (shuffling)



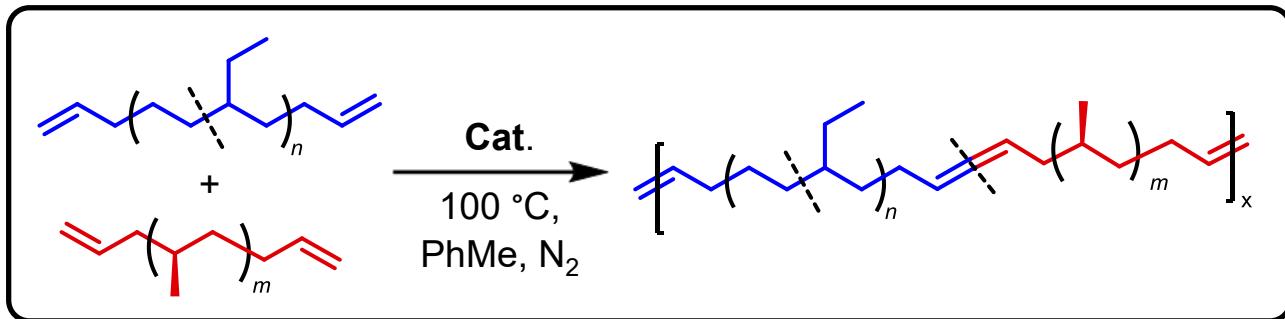
- Homogeneous phase
- High crystallinity / modulus
- Compatibilizer

TEM images of Dow materials from:

Munro, J.; Hu, Y.; Laakso, R.; Madenjian, L.; Vervoort, S.; Werner, S.; Marchand, G.  
Polypropylene-Rich Blends with Ethylene/α-Olefin Copolymers Compatibilized with INTUNE™ Propylene-Olefin Block Copolymers.

Infuse chemistry: Arriola, D. J.; Carnahan, E. M.; Hustad, P. D.; Kuhlman, R. L.; Wenzel, T. T., *Science* **2006**, 312, 714–719.

# Shuffling Catalyst Selection



Average number of blocks:

(3 mol% Ru catalyst)

1.9

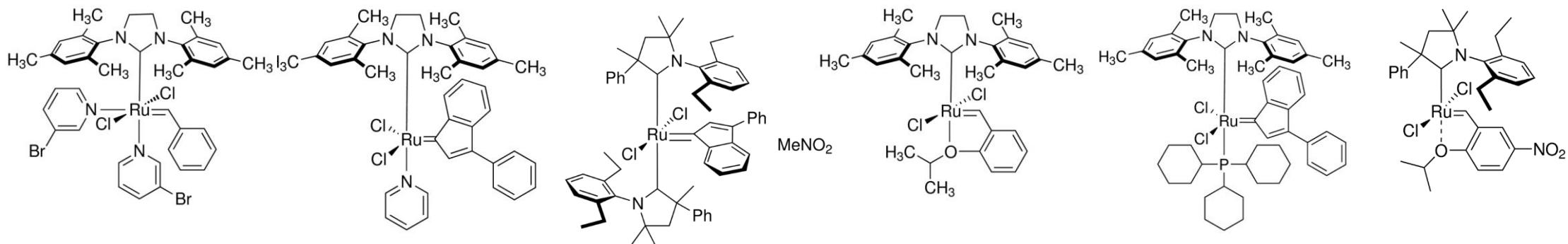
3.2

3.3

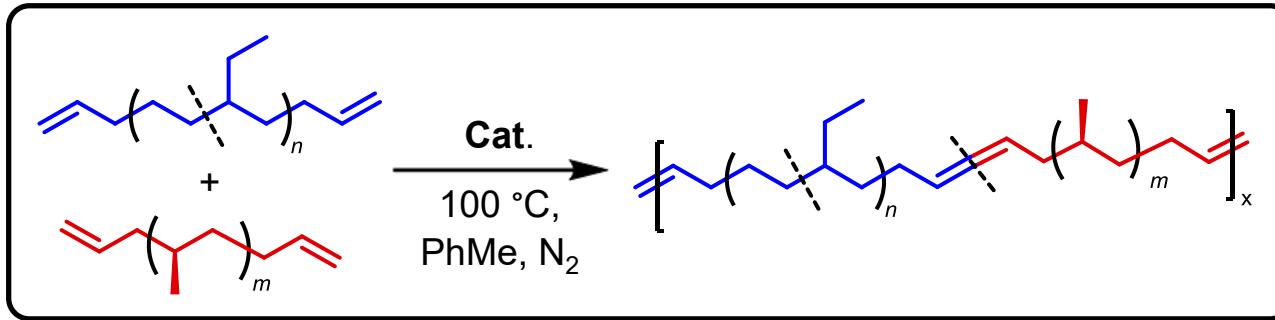
4.0

5.2

5.8

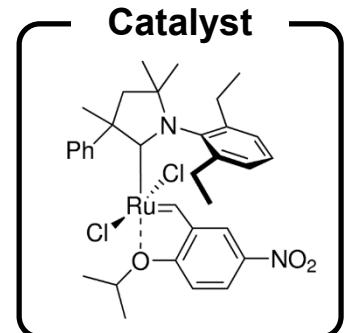
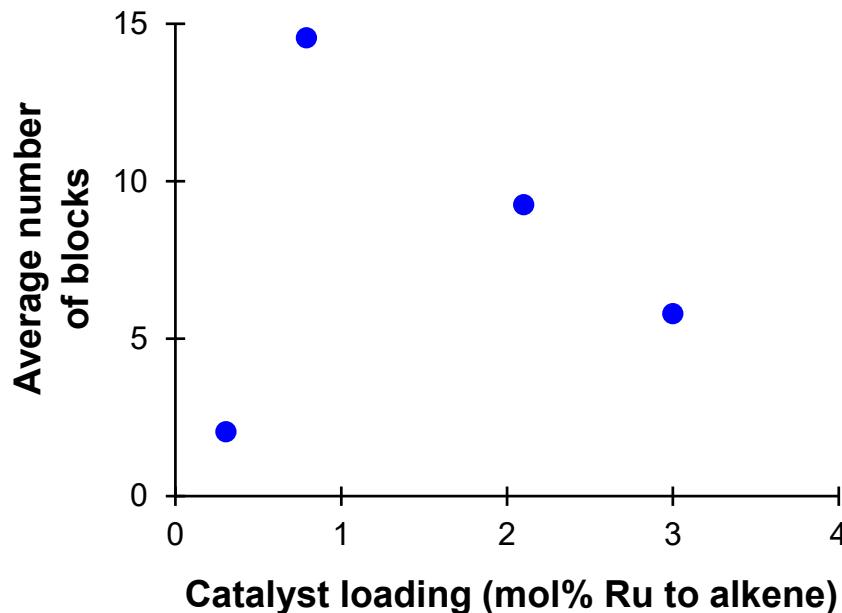


# Shuffling Catalyst Selection



## Key Catalyst Findings:

- Bulky cyclic alkyl amino carbenes improve thermal and ethylene stability
- Monofunctional Ru alkylidene limits molecular weight
- Sufficient catalyst activity is required for reactivity

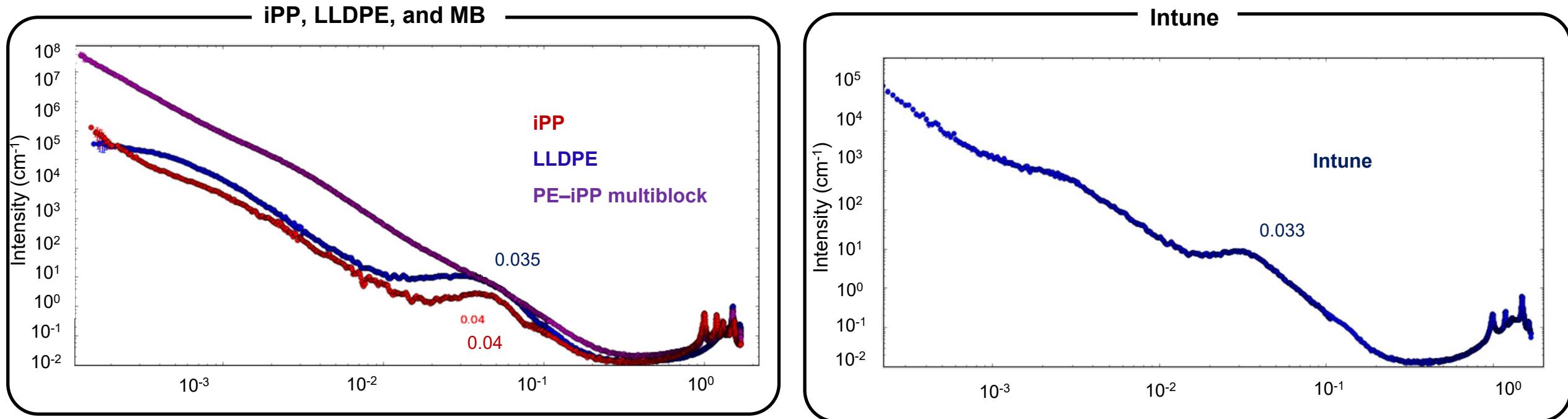


# $T_1$ Relaxation Studies

Sample	$T_1$ (ms)	
	PP	PE
iPP/PE (Blend)	792	1921
Intune (Diblock)	658	667
iPP–PE (Multiblock) (Sample 1)	765	851

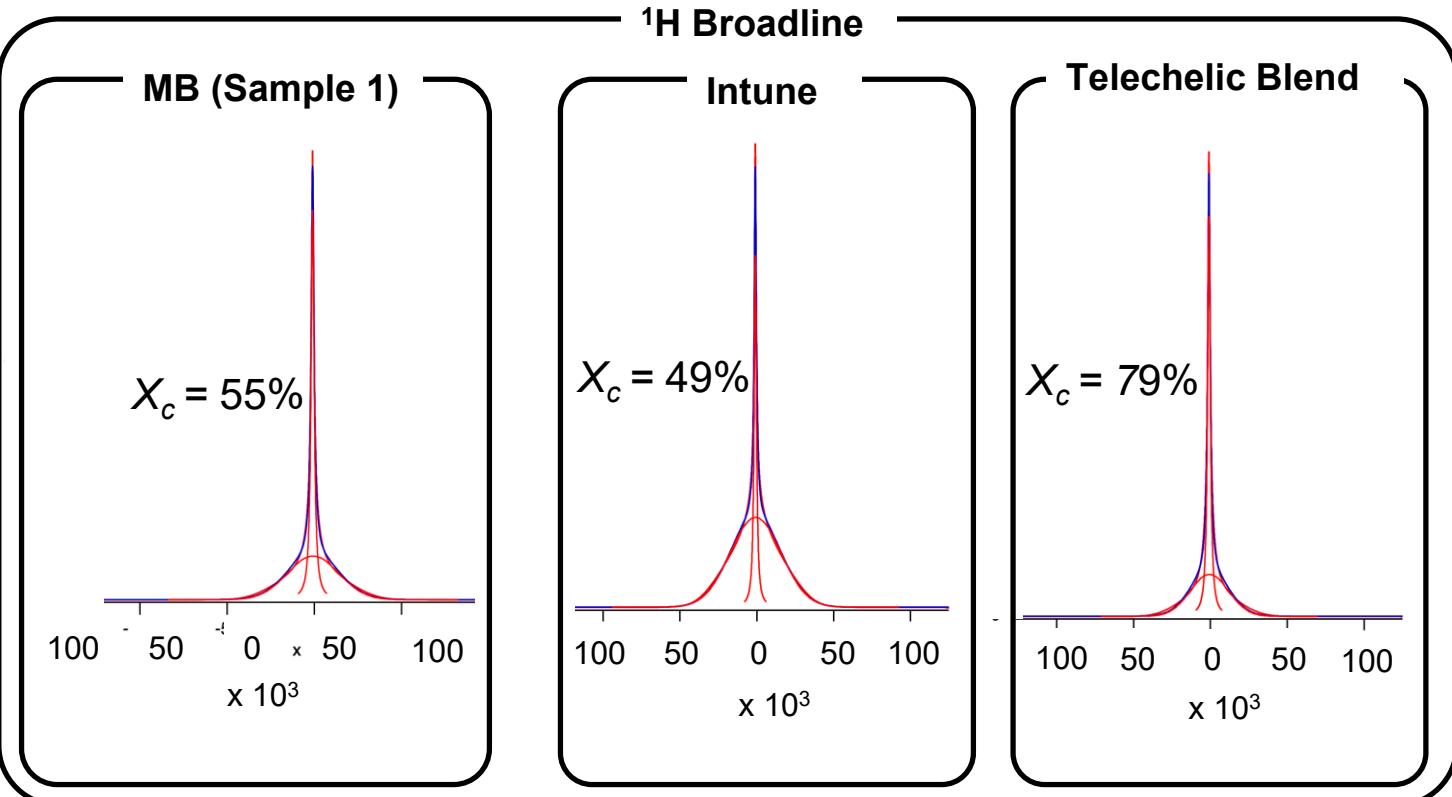
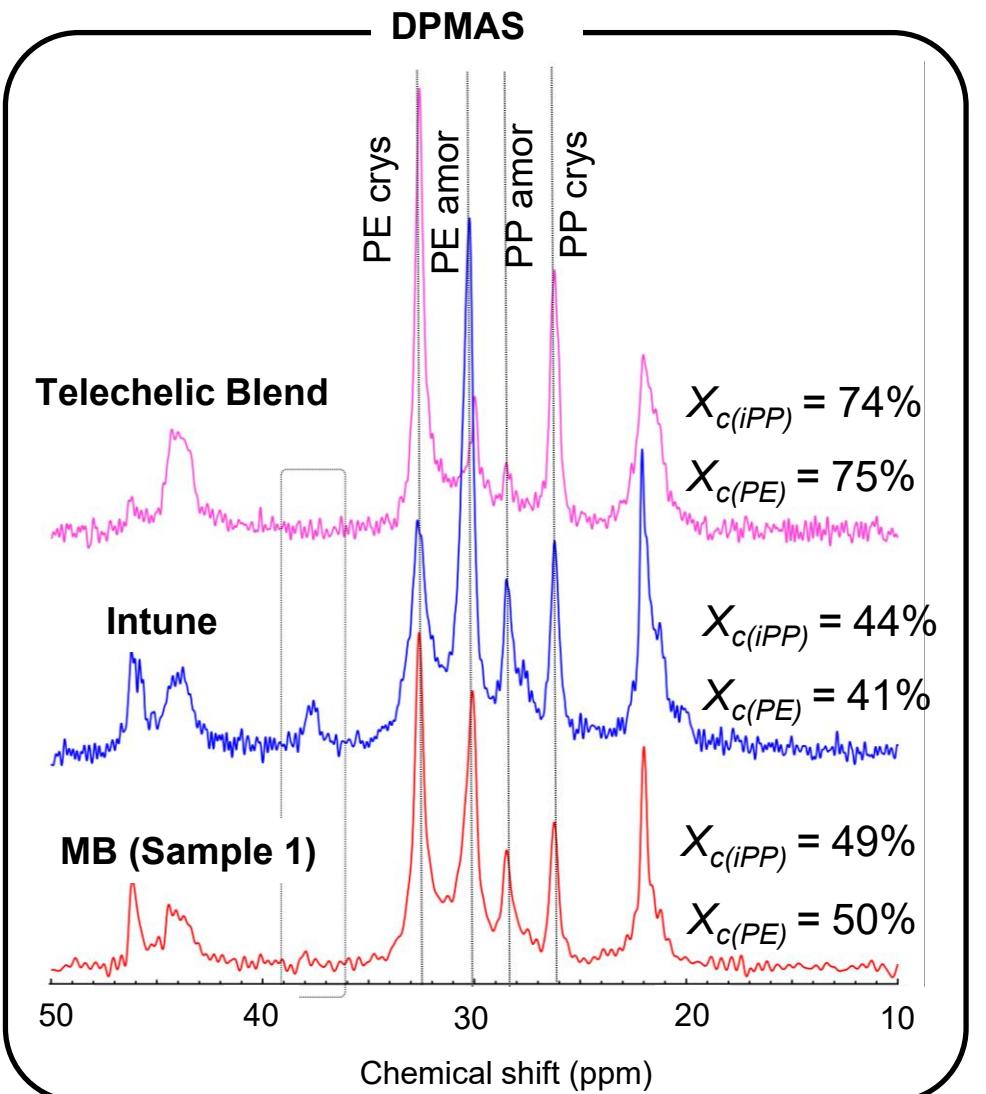
Covalently linked segments exhibit similar relaxation times

# SAXS of Multiblock Product

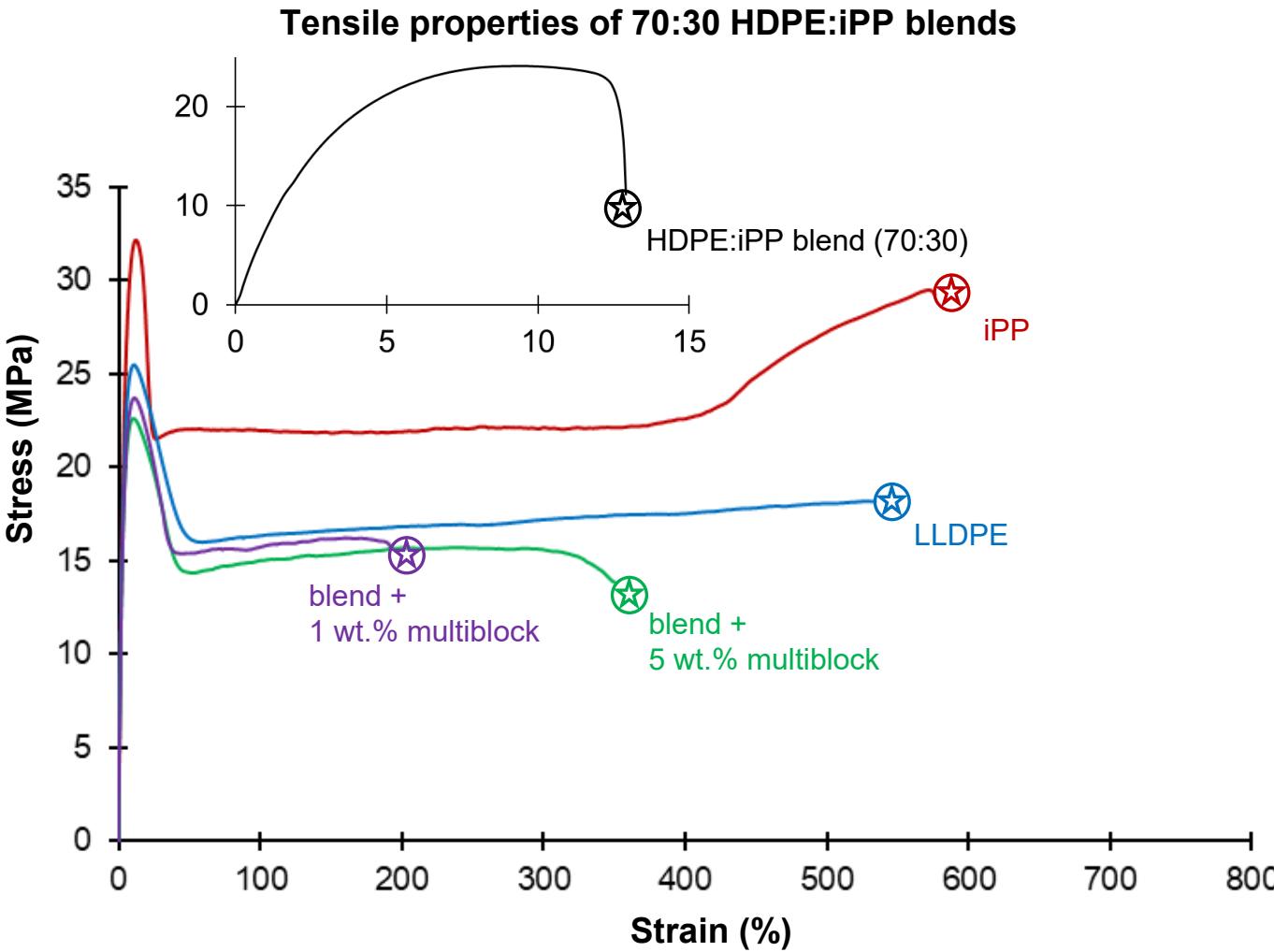


- Scattering invariant increases for the multiblock sample.
- Greater number of domains leads to more interfaces, which leads to more scattering intensity

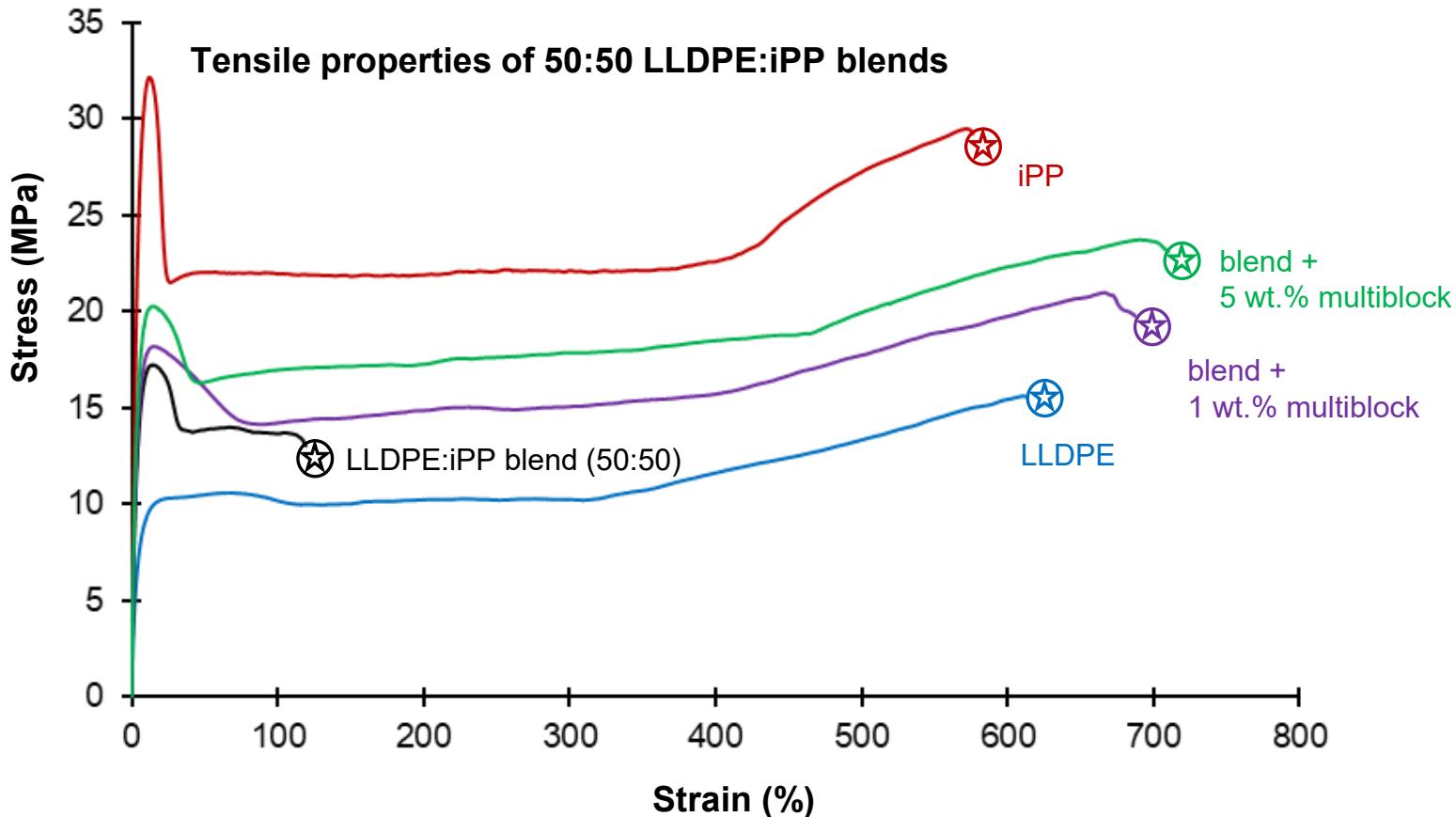
# Crystallinity Studies of Multiblock Product



# Compatibilization Efficiency

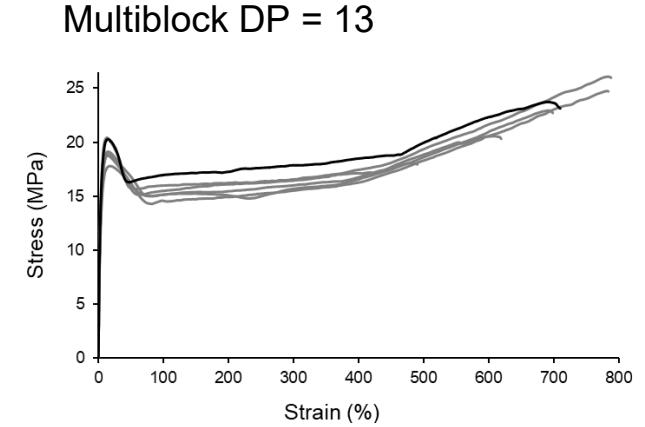
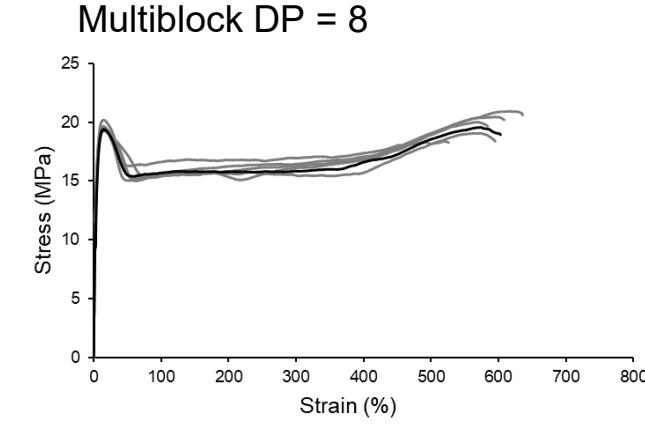
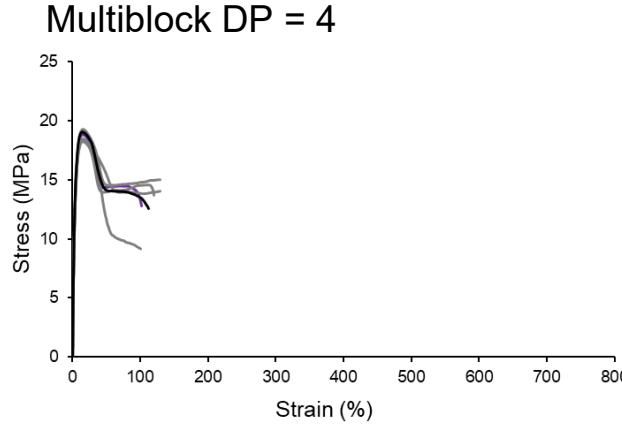


# Compatibilization Efficiency

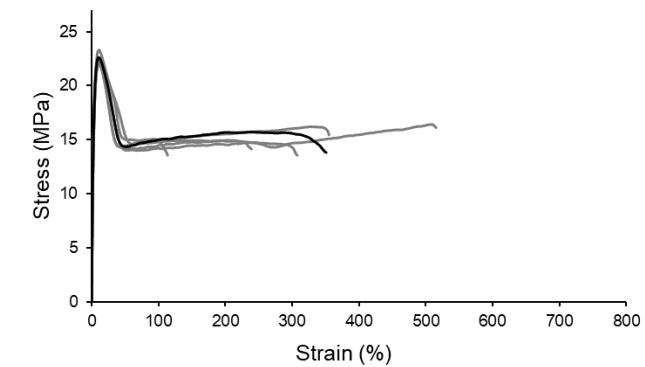
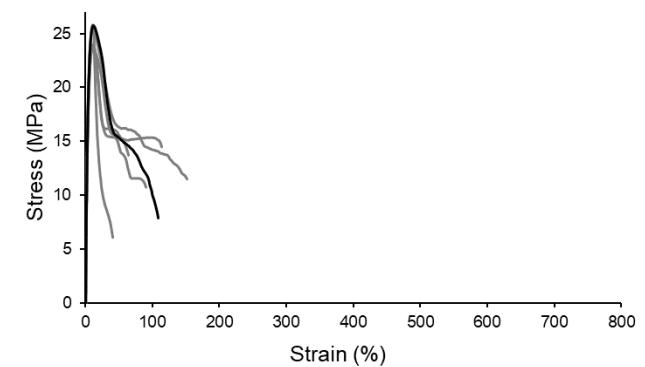
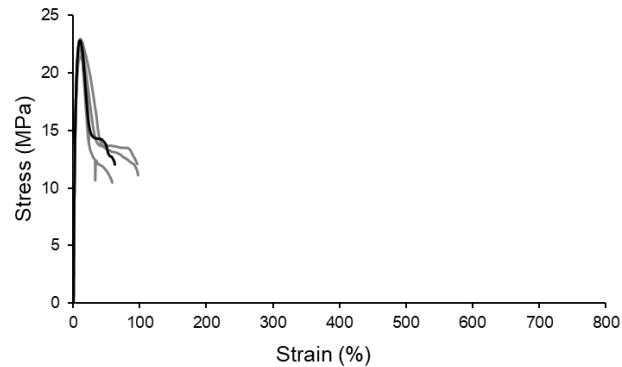


# Molecular Weight Effects of MBCP

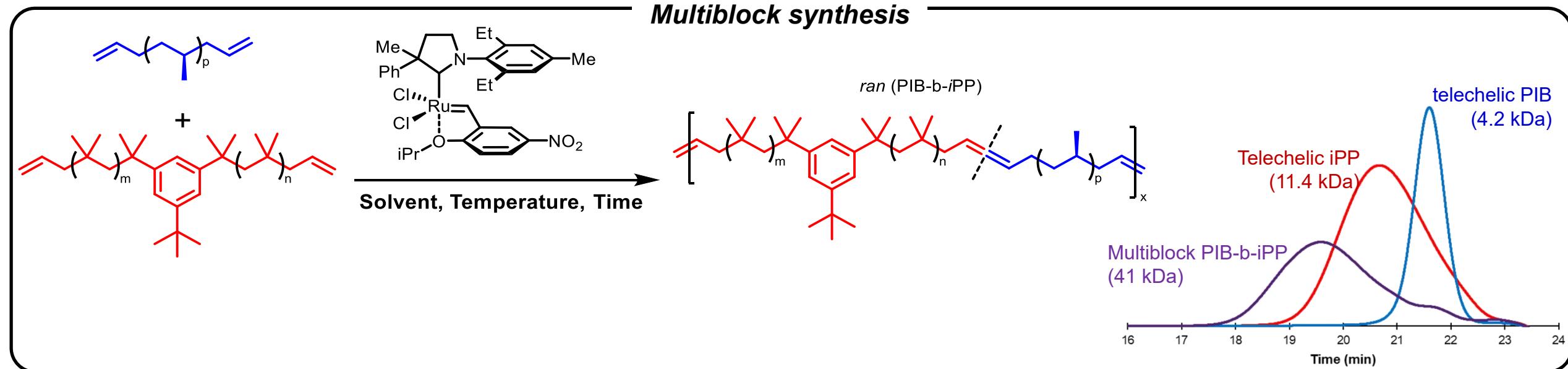
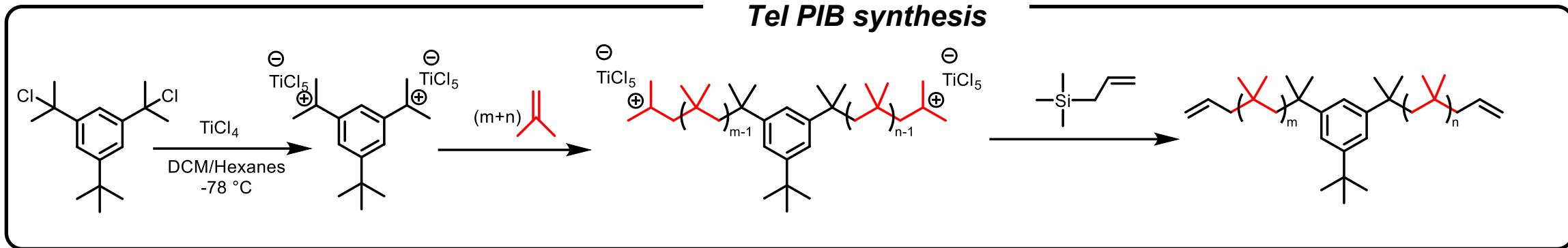
50:50 LLDPE:iPP blends  
+ 5% compatibilizer



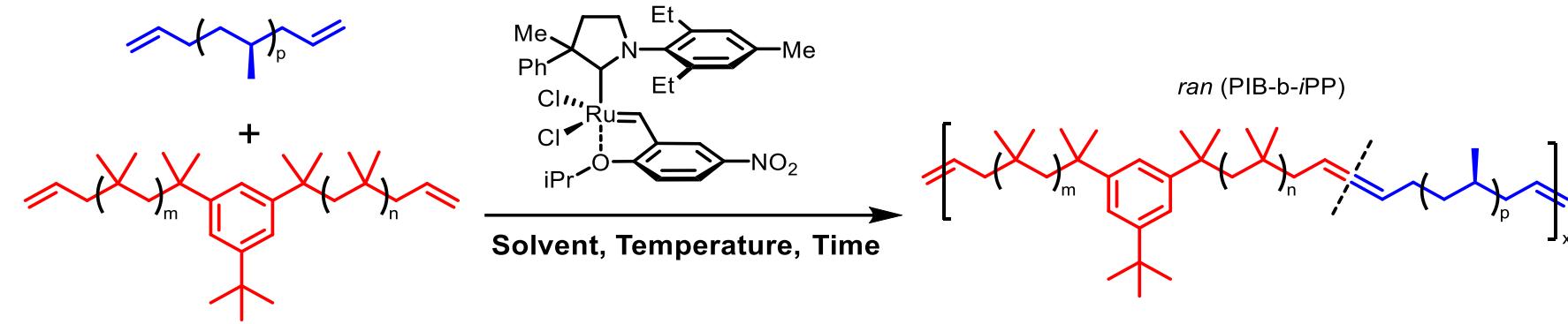
70:30 HDPE:iPP blends  
+ 5% compatibilizer



# Future Work (In Progress)

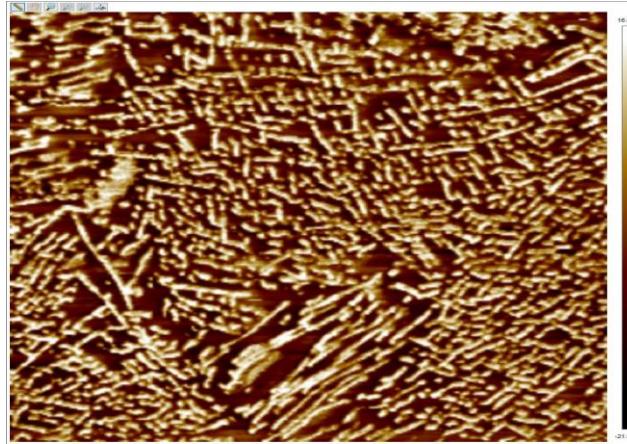
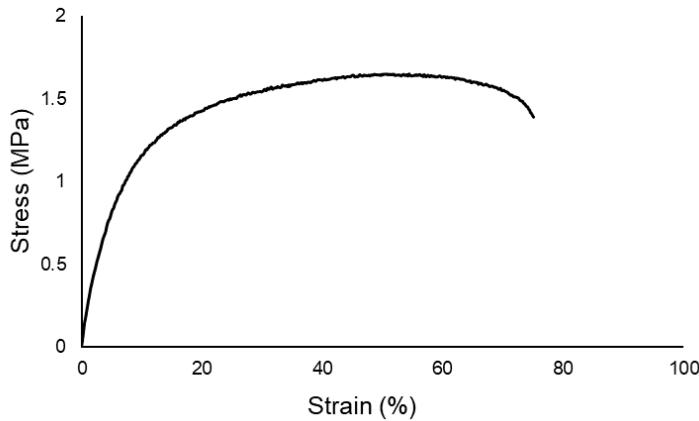


# Future Work (In Progress)



Yuliana  
Ospina-Yepes

Baseline multiblock properties



# Acknowledgements



**Funding & Resources Provided By:**



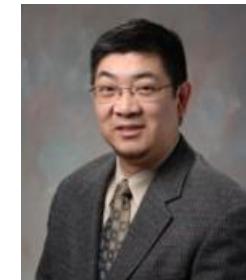
**Advisor:**



**Collaborators:**



Aarushi Srivastava    Navin Kafle



Dr. Li Jia



Dr. Tohi Miyoshi



**Materials and Samples:**

